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A novel chemical synthesis of a 3-deoxy-D-*arabino*-heptulosonic acid 7-phosphate (DAHP) derivative and its 2-deoxy analogue

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Abstract

Using 2,3,4,6-tetra-*O*-acetyl-β-D-glucopyranosyl cyanide as a precursor, methyl (methyl 3-de-oxy-α-D-*arabino*-hept-2-ulopyranosid)onate (6) and its 2-deoxy analogue (10) were prepared. The synthesis involved an elimination of one molecule of acetic acid from C-2–C-3 and transformation of the CN group into COOMe, followed by methoxymercuration with subsequent reductive removal of the mercuri residue to give 6 or hydrogenation of the double bond to give 10. Phosphorylation of the 7-OH group led to the title compounds. © 1996 Elsevier Science Ltd.

Keywords: DAHP; D-arabino-Heptulosonic acid; 2-Deoxy-D-arabino-heptulosonic acid

1. Introduction

Its well known that all micro-organisms and plants synthesise the aromatic amino acids (phenylalanine, tyrosine, tryptophan) from 3-deoxy-D-arabino-hept-2-ulosonic acid 7-phosphate (DAHP) by enzyme-catalysed transformations (shikimic pathway) [1]. To investigate the mechanism of this transformation the synthetic compounds were required. Two approaches, chemical [2] and enzymatic [3], for the synthesis of DAHP have been reported. The chemical syntheses of this compound are usually based on 2-deoxy-D-arabino-hexose diethyl dithioacetal [2].

A more general approach to the ulosonic acids has been demonstrated by Crich and Ritchie, who used glycals for the C-1 carbon elongation [4]. In an anionic-free-radical process, utilising the reactivity of the C-1 phenyl sulphones, they introduced COOH and

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OH groups at C-1 of the pyranoses. The present paper describes an addition reaction to the 1-cyanoglucal derivative as a route to the title compounds.

2. Results and discussion

As a continuation of our studies aimed at the synthesis of 3-deoxy-2-ulosonic acids, we recently reported [5] the use of 1-cyanoglycals [6,7] as masked synthons for carboxyl and hydroxyl functions. This approach seemed to be especially suitable for the synthesis of 3-deoxy-D-*arabino*-heptulosonic acid 7-phosphate (DAHP), starting from 1-cyano-D-glucal, which contains all the structural and stereochemical features of the natural DAHP, except those at the anomeric centre and at C-7.

Slightly modifying a previously elaborated procedure for the elimination of one molecule of acetic acid from glycopyranosyl cyanides [7], the D-gluco derivative 1 was treated with 1,8-diazabicyclo[5.4.0]undec-7ene (DBU), leading to 4,5,7-tri-O-acetyl-2,6-anhydro-3-deoxy-D-arabino-hept-2-enononitrile (2) in a 70% yield (Scheme 1). Previously, compound 2 has been obtained by Buchanan et al. from 2,3,4,6-tetra-O-acetyl- β -D-mannopyranosyl cyanide, using a similar procedure [8]. Although the sub-

Scheme 1. (a) DBU, CH_2Cl_2 ; (b) 1. $NaOH_{aq}$ -EtOH, 2. Ac_2O -Py, 3. CH_2N_2 -Et $_2O$; (c) 1. $Hg(CF_3COO)_2$ -MeOH, 2. KCl; (d) Ph_3SnH , toluene; (e) Me_3SiCl , MeOH, or $NaHCO_3$, MeOH; (f) $(PhO)_2POCl$ -Py; (g) PtO_2 , H_2 , EtOH; (h) Pd-C, H_2 , EtOH.

strate described [8], having the *trans*-diaxial arrangement of H-1 and AcO-2, should be more favourable for an elimination of acetic acid, there is a precedent in the literature which suggests that the stereochemistry at C-1-C-2 is not crucial for this process [9].

Hydrolysis of **2** with 1 M aqueous ethanolic NaOH at ~ 90 °C, then acetylation, and esterification of the carboxyl group, using diazomethane, afforded methyl 4,5,7-tri-O-acetyl-2,6-anhydro-3-deoxy-D-arabino-hept-2-enonate (3) in 91% yield.

Glycosidation of the glycal 3 was promoted by a mercuric salt. Thus, treatment of 3 with mercuric trifluoroacetate in methanol followed by potassium chloride gave a mixture of two diastereomers of α -D-manno (4a) and β -D-gluco (4b) configuration. To provide a proof of the stereochemistry at C-2 of 4a and 4b, the isomers were separated by flash chomatography on silica gel. Their structures were established by ¹H NMR data. The 500-MHz spectrum of the major product 4a (the ratio of 4a:4b = 4:1) exhibited a well-defined doublet at δ 3.36 for H-3 with a $J_{3,4}$ value of 5.2 Hz, supporting a cis axial-equatorial arrangement between the 3-chloromercuri substituent and the adjacent 4-acetoxy group. In contrast, the upfield shift to δ 2.53 (doublet) for H-3 in 4b, as well as the $J_{3,4}$ value of 12.0 Hz, consistent with a trans equatorial-equatorial arrangement of the 3-chloromercuri and 4-acetoxy substituents, confirmed unambiguously the β -anomeric configuration of 4b. A prevalence of the trans-diaxial addition product 4a over the trans-diequatorial 4b is comparable to previous findings on glycosidation of 1-methoxycarbonylglycals (sialic acid series) with N-bromo- and N-iodo-succinimide in methanol [10].

Reductive demercuration of the individual isomers $\bf 4a$ and $\bf 4b$ with triphenyltin hydride in the presence of an excess of NaOAc [11] resulted in the formation of methyl (methyl 4,5,7-tri-O-acetyl-3-deoxy- α -D-arabino-hept-2-ulopyranosid)onate ($\bf 5a$, DAH derivative) and its β anomer $\bf 5b$, respectively. The configuration of the products $\bf 5a$ and $\bf 5b$ was a final proof for trans-diaxial and trans-diequatorial addition of the methoxy-chloromercuri groups across the double bond in $\bf 3$.

It is worth noting that DAH accompanies DAHP produced by different strains of *Escherichia coli*. Assuming, however, that DAHP is formed from phosphoenol pyruvate and D-erythrose 4-phosphate in a condensation mediated by DAHP synthase, DHA may rather be a product of dephosphorylation of DAHP [2].

To introduce the phosphate group at O-7, deacetylation of 5, leaving the C-1 methoxycarbonyl group untouched, was necessary. It was done by treatment of a methanolic solution of 5 with dry NaHCO₃ at room temperature over a 1-h period; the desired compound 6 was then isolated in high yield as the sole product. This sensitivity of the acetyl residues in 6 towards NaHCO₃ was very useful. Similar sensitivity of these acetates in 5 and 9 was observed in the presence of chlorotrimethylsilane in methanol to give 6 and 10, respectively.

Further steps, leading to the DAHP derivative, were as follows: phosphorylation of **6a** with diphenyl phosphonate in pyridine [2] at -20 °C followed by flash chromatography on silica gel afforded the 7-phosphate derivative **7** in 95% yield. Diphenyl ester groups were removed by hydrogenolysis using Adams' catalyst, to give the phosphate **8**, isolated as a very hygroscopic white solid.

Methyl 4,5,7-tri-O-acetyl-2,6-anhydro-3-deoxy-D-arabino-hept-2-enonate (3), due to the unsaturation, is a potential precursor of modified heptulosonic acids. The simplest

modification that we performed was the hydrogenation of 3. Compound 9 thus obtained, being a 2-deoxy analogue of DAH, is a known molecule [12]. By the route described above for DAHP synthesis, we prepared its 7-phosphate analogue 12.

In summary, a convenient synthon for both DAH and DAHP, as well as for their 2-deoxy analogues, was prepared. Further investigations into the utilisation of 1-cyanoglycals in the synthesis of the 3-deoxy-2-ulosonic acids are now in progress.

3. Experimental

General methods.—Optical rotations were measured with a JASCO DIP Digital Polarimeter at room temperature. ¹H NMR spectra were recorded on Bruker AM-500 (500 MHz) and Varian AC-200 (200 MHz) spectrometers with Me₄Si as internal standard. Mass spectra were taken on an AMD-604 mass spectrometer. Reactions were monitored by TLC on silica gel [Merck alu-plates (0.2 mm)]. All reagents and solvents were purified and dried according to common methods [12]. Pyridine (used for phosphorylation reactions) was dried by distillation over KOH, CaH₂, and TsCl, successively, and left over CaH₂ in a dark glass bottle with serum cap. All organic solutions were dried over MgSO₄. Reaction products were purified by flash column chromatography, using Merck Kieselgel 60 (240–400 or 70–230 mesh).

4,5,7-Tri-O-acetyl-2,6-anhydro-3-deoxy-D-arabino-hept-2-enononitrile (2).—To a stirred solution of cyanide 1 (2.0 g, 5.60 mmol) [6] in dry CH₂Cl₂ (50 mL) was added DBU (2.0 mL, 13.3 mmol) at -20 °C. Stirring was continued at this temperature for 5 h, and the mixture was left overnight at -5 °C. TLC (1:1 hexane-ether) showed that some starting material remained. Since a prolonged reaction time or a higher temperature led to the destruction of the substrate, the reaction was interrupted and, after processing, the mixture was separated by chromatography on silica gel. Elution with 2:3 hexane-ether removed the unchanged 1, and gave the desired 2 (1.2 g, 70%); mp 78–80 °C; $[\alpha]_D - 42.4^\circ$ (c 2.0, CHCl₃); lit. [8] mp 79–81 °C, $[\alpha]_D - 46.6^\circ$ (CHCl₃); ¹H NMR (200 MHz, CDCl₃): δ 2.08, 2.10, 2.12 (3 s, 9 H, 3 × OAc), 4.19 (m, 1 H, H-7a), 4.38–4.50 (m, 2 H, H-6, H-7b), 5.22 (dt, 1 H, H-5), 5.38 (dd, 1 H, H-4), 5.72 (d, 1 H, H-3); $J_{3,4}$ 3.7, $J_{4,5}$ 5.5, $J_{5,6}$ 5.8 Hz, $J_{6,7a}$, $J_{6,7b}$, and $J_{7a,7b}$ - unresolved. Anal. Calcd for C₁₃H₁₅NO₇: C, 52.53; H, 5.09; N, 4.71. Found: C, 52.34; H, 5.11; N, 4.73.

Methyl 4,5,7-tri-O-acetyl-2,6-anhydro-3-deoxy-D-arabino-hept-2-enonate (3).—To a solution of nitrile 2 (3.0 g, 10.0 mmol) in EtOH (100 mL) was added 1 M aq NaOH (80 mL). The reaction mixture was heated to ~90 °C for 20 h, then cooled to room temperature and neutralised with AcOH. After evaporation under reduced pressure, the solid was treated with an excess of 1:1 Ac₂O-pyridine for 12 h. The mixture was then poured into ice—water, acidified with aq HCl to pH ~ 3, and extracted with EtOAc. The organic layers were washed with water, dried, and concentrated. The resulting syrup was redissolved in MeOH and treated with a solution of CH₂N₂ in ether. Filtration through a short column of silica gel (7:3 hexane–acetone) gave 3 (3.0 g, 91%) as a colourless oil; $[\alpha]_D$ – 55.7° (c 0.69, CHCl₃); ¹H NMR (500 MHz, CDCl₃): δ 2.07, 2.08, 2.10 (3 s, 9 H, 3 × OAc), 3.83 (s, 3 H, COOMe), 4.22 (dd, 1 H, H-7a), 4.44 (m, 1 H, H-6), 4.47 (t, 1 H, H-7b), 5.25 (dd, 1 H, H-5), 5.44 (m, 1 H, H-4), 6.04 (d, 1 H, H-3); $J_{3,4}$ 3.7, $J_{3,5}$

0.3, $J_{4,5}$ 5.4, $J_{4,6}$ 0.8, $J_{5,6}$ 6.7, $J_{6,7a}$ 2.8, $J_{6,7b}$ 5.8, $J_{7a,7b}$ 11.8 Hz. Anal. Calcd for $C_{14}H_{18}O_9$: C, 50.91; H, 5.49. Found: C, 51.20; H, 5.62.

Methyl (methyl 4,5,7-tri-O-acetyl-3-chloromercuri-3-deoxy-α-D-manno- (4a) and methyl (methyl 4,5,7-tri-O-acetyl-3-chloromercuri-3-deoxy-β-D-gluco-hept-2-ulopyranosid)onate (4b).—To a stirred solution of ester 3 (3.0 g, 9.08 mmol) in MeOH (60 mL) was added stepwise mercuric trifluoroacetate (7.8 g, 18.16 mmol) at 0 °C under Ar. Stirring was continued at room temperature. After 0.5 h, when TLC (3:2 hexane-acetone) showed that 3 was all consumed, KCl (2 g) was added, and the mixture was left overnight. Column chromatography of the product (1:3 hexane-ether) gave pure two diastereoisomers in a ratio $\alpha:\beta=4:1$ (overall yield 80%). Eluted first was the β -D-gluco isomer (4b) (16%); mp 61–63°C; $[\alpha]_D$ +75.3° (c 1.7, CHCl₃); ¹H NMR (200 MHz, CDCl₃): δ 2.03, 2.07, 2.10 (3 s, 9 H, 3 × OAc), 2.53 (dd, 1 H, H-3), 3.37 (s, 3 H, OMe), 3.89 (ddd, 1 H, H-6), 3.93 (s, 3 H, COOMe), 4.13-4.35 (m, 2 H, H-7, H-7a), 4.93–5.15 (m, 2 H, H-4, H-5); $J_{3,4}$ 12.0, $J_{4,5}$ 9.2, $J_{5,6}$ 9.9, $J_{6,7}$ 2.5, $J_{6,7a}$ 4.3, $J_{7,7a}$ 12.4 Hz. Eluted second was the α -D-manno isomer (4a) (64%); mp 145 °C; [α]_D -25.2° (c1.75, CHCl₃); ¹H NMR (200 MHz, CDCl₃): δ 2.05, 2.06, 2.14 (3 s, 9 H, 3 × OAc), 3.33 (s, 3 H, OMe), 3.36 (d, 1 H, H-3), 3.88 (s, 3 H, COOMe), 3.98 (ddd, 1 H, H-6), 4.18 (dd, 1 H, H-7a), 4.28 (dd, 1 H, H-7b), 4.98 (dd, 1 H, H-5), 5.81 (dd, 1 H, H-4); J_{3.4} 5.2, $J_{4,5}$ 8.7, $J_{5,6}$ 9.9, $J_{6,7a}$ 2.5, $J_{6,7b}$ 4.3, $J_{7a,7b}$ 12.4 Hz.

Methyl (*methyl* 4,5,7-tri-O-acetyl-3-deoxy-α-D-arabino-hept-2-ulopyranosid)onate (5a).—A mixture of 4a (3.2 g, 5.36 mmol) and dry NaOAc (2.1 g, 25 mmol) in dry toluene (50 mL) was treated with triphenyltin hydride (solution in toluene) (2.1 g, 6.0 mmol) and azobis(isobutyronitrile) (10 mg) at 0 °C under Ar. The reaction was stirred overnight; TLC (1:3 hexane–ether) showed that 4a was all consumed. Evaporation left a residue which was dissolved in ether, treated with aq 30% KF, and stirred overnight. After filtration through Celite, the organic phase was separated, dried, and evaporated. Column chromatography of the residue (2:3 hexane–ether) gave 5a as a colourless oil (1.7 g, 90%); $[\alpha]_D$ +46.4° (*c* 1.04, CHCl₃); ¹H NMR (500 MHz, CDCl₃): δ 1.87 (dd, 1 H, H-3ax), 2.01, 2.05, 2.09 (3 s, 9 H, 3 × OAc), 2.52 (dd, 1 H, H-3eq), 3.29 (s, 3 H, OMe), 3.81 (s, 3 H, COOMe), 3.90 (ddd, 1 H, H-6), 4.15 (dd, 1 H, H-7a), 4.32 (dd, 1 H, H-7b), 5.04 (t, 1 H, H-5), 5.34 (m, 1 H, H-4); $J_{3ax,3eq}$ 13.0, $J_{3ax,4}$ 11.5, $J_{3eq,4}$ 5.4, $J_{4,5}$ 9.7, $J_{5,6}$ 10.0, $J_{6,7a}$ 2.4, $J_{6,7b}$ 5.2, $J_{7a,7b}$ 12.3 Hz. Anal. Calcd for $C_{15}H_{22}O_{10}$: C, 49.72; H, 6.12. Found: C, 49.52; H, 6.30.

Methyl (*methyl* 4,5,7-tri-O-acetyl-3-deoxy-β-D-arabino-hept-2-ulopyranosid)onate (**5b**).—Processing was similar to that described for the preparation of **5a**, to give **5b** as a colourless oil (89%); $[\alpha]_D$ +46.5° (c 1.49, CHCl $_3$); H NMR (500 MHz, CDCl $_3$): δ 1.94 (dd, 1 H, H-3ax), 2.03, 2.04 2.10 (3 s, 9 H, 3 × OAc), 2.65 (dd, 1 H, H-3eq), 3.36 (s, 3 H, OMe), 3.86 (s, 3 H, COOMe), 4.01 (ddd, 1 H, H-6), 4.20 (dd, 1 H, H-7a), 4.31 (dd, 1 H, H-7b), 4.99 (m, 1 H, H-4), 5.05 (t, 1 H, H-5); $J_{3ax,3eq}$ 13.2, $J_{3ax,4}$ 11.1, $J_{3eq,4}$ 5.0, $J_{4,5}$ 9.2, $J_{5,6}$ 9.4, $J_{6,7a}$ 2.6, $J_{6,7b}$ 4.4, $J_{7a,7b}$ 12.3 Hz. HR-MS (LSIMS): Calcd for C $_{15}$ H $_{22}$ O $_{10}$ [M + Na]: m/z 385.1111. Found: m/z 385.1116.

Methyl (methyl 3-deoxy-α-D-arabino-hept-2-ulopyranosid)onate (6a).—To a solution of ester 5a (2.0 g, 5.52 mmol) in MeOH (10 mL) was added NaHCO₃ (2 g) and the resulting suspension was stirred at room temperature for 1 h (TLC, 9:1 CH₂Cl₂-MeOH). The mixture was filtered through Celite and a short column of silica gel to give 6a

(white crystals, 1.2 g, 90%); mp 146–147 °C; $[\alpha]_D$ +79.4° (c 0.90, MeOH); ¹H NMR (500 MHz, D₂O): δ 1.75 (dd, 1 H, H-3ax), 2.36 (dd, 1 H, H-3eq), 3.25 (s, 3 H, OMe), 3.42 (t, 1 H, H-5), 3.59 (ddd, 1 H, H-6), 3.84 (dd, 1 H, H-7b), 3.87 (s, 3 H, COOMe), 3.91–3.98 (m, 2 H, H-4, H-7a); $J_{3ax,3eq}$ 13.4, $J_{3ax,4}$ 11.6, $J_{3eq,4}$ 5.2, $J_{4,5}$ 9.5, $J_{5,6}$ 10.0, $J_{6,7a}$ 2.3, $J_{6,7b}$ 5.5, $J_{7a,7b}$ 12.4 Hz.

Methyl (*methyl* 3-deoxy-β-D-arabino-hept-2-ulopyranosid)onate (**6b**).—Processing was similar to that described for the preparation of **6a**, to give **6b** as a colourless oil (95%); $[\alpha]_D$ +44.3° (*c* 0.89, MeOH); ¹H NMR (500 MHz, D₂O): δ 1.63 (dd, 1 H, H-3ax), 2.45 (dd, 1 H, H-3eq), 3.22 (s, 3 H, OMe), 3.25 (t, 1 H, H-5), 3.40–350 (m, 2 H, H-4, H-6), 3.64 (dd, 1 H, H-7b), 3.72 (s, 3 H, COOMe), 3.76 (dd, 1 H, H-7a); $J_{3ax.3eq}$ 12.9, $J_{3ax.4}$ 11.9, $J_{3eq.4}$ 4.8, $J_{4.5}$ 9.2, $J_{5.6}$ 9.7, $J_{6.7a}$ 2.3, $J_{6.7b}$ 5.0, $J_{7a.7b}$ 12.4 Hz.

Methyl (methyl 3-deoxy-α-D-arabino-hept-2-ulopyranosid)onate 7-(diphenyl phosphate) (7).—The ester **6a** (100 mg, 0.42 mmol) was dissolved in dry pyridine (2 mL) and freshly distilled diphenyl phosphorochloridate (0.1 mL, 0.50 mmol) was added at -20 °C under Ar. The reaction mixture was stirred until all substrate **6a** was consumed (20–30 min) (TLC, EtOAc). The reaction mixture was then diluted with toluene and concentrated under reduced pressure. Pyridine was removed by evaporation with toluene. Flash chromatography (95:5 CH₂Cl₂–MeOH) of the mixture on silica gel yielded **7** (190 mg, 95%) as a light-yellow oil; [α]_D +14.5° (c 2.52, CHCl₃); ¹H NMR (500 MHz, CDCl₃): δ 1.58 (dd, 1 H, H-3ax), 2.34 (dd, 1 H, H-3eq), 3.19 (s, 3 H, OMe), 3.24 (t, 1 H, H-5), 3.56 (m, 1 H, H-6), 3.81 (s, 3 H, COOMe), 3.94 (m, 1 H, H-4), 4.43 (td, 1 H, H-7a) 4.63 (m, 1 H, H-7b), 7.20–7.37 (m, 10 H, Ar); $J_{3ax,3eq}$ 13.1, $J_{3ax,4}$ 11.4, $J_{3eq,4}$ 5.1, $J_{4,5}$ 9.2, $J_{5,6}$ 9.7, $J_{6,7a}$ 2.0, $J_{6,7b}$ 3.5, $J_{7a,7b}$ 11.9, $J_{7a,p}$ 12.0, $J_{7b,p}$ 9.8 Hz.

Methyl (methyl 3-deoxy-α-D-arabino-hept-2-ulopyranosid)onate 7-(dihydrogen phosphate) (8).—The phosphate ester 7 (100 mg, 0.21 mmol) dissolved in EtOH (96%, 6 mL) was vigorously stirred with PtO₂ (50 mg) under H₂ for 12 h. The reaction mixture was filtered through Celite, and the filtrate was concentrated. The residue was washed with ether to give 8 as a white, very hygroscopic solid (65 mg, 100%); $[\alpha]_D + 51.5^\circ$ (c 1.20, MeOH); ¹H NMR (500 MHz, D₂O): δ 1.77 (dd, 1 H, H-3ax), 2.35 (dd, 1 H, H-3eq), 3.25 (s, 3 H, OMe), 3.49 (t, 1 H, H-5), 3.70 (bdd, 1 H, H-6), 3.85 (s, 3 H, COOMe), 3.94 (m, 1 H, H-4), 4.12–4.22 (m, 2 H, H-7a, H-7b); $J_{3ax,3eq}$ 13.3, $J_{3ax,4}$ 11.6, $J_{3eq,4}$ 5.2, $J_{4,5}$ 9.2, $J_{5,6}$ 9.5, $J_{6,7a}$ 2.0, $J_{6,7b}$ 5.7, $J_{7a,7b}$ 11.7, $J_{7a,P}$ 5.7, $J_{7b,P}$ 7.0 Hz. HR-MS (LSIMS): Calcd for $C_9H_{17}O_{10}P$ [M + Na]: m/z 339.0457. Found: m/z 339.0453.

Methyl 4,5,7-tri-O-acetyl-2,6-anhydro-3-deoxy-D-gluco-heptonate (9).—The ester 3 (250 mg, 0.75 mmol) dissolved in EtOH (5 mL) was vigorously stirred with Pd–C (200 mg) under H₂ for 12 h. TLC (3:2 hexane–EtOAc) showed disappearance of the substrate. The mixture was filtered through Celite, and the filtrate was concentrated, to give 9 (250 mg, 100%) as a colourless oil; $[\alpha]_D - 0.8^\circ$ (c 1.05, CHCl₃); ¹H NMR (500 MHz, CDCl₃): δ 1.63, 1.68, 1.69 (3 s, 9 H, 3 × OAc), 1.66 (dd, 1 H, H-3ax), 2.20 (ddd, 1 H, H-3eq), 3.15 (ddd, 1 H, H-6), 3.25 (s, 3 H, COOMe), 3.54 (dd, 1 H, H-2), 4.07 (dd, 1 H, H-7a), 4.33 (dd, 1 H, H-7b), 4.99 (m, 1 H, H-4), 5.18 (t, 1 H, H-5); $J_{2.3ax}$ 12.1, $J_{2.3eq}$ 2.2, $J_{3ax,3eq}$ 12.9, $J_{3ax,4}$ 9.5, $J_{3eq,4}$ 5.3, $J_{4.5}$ 9.8, $J_{5.6}$ 10.2, $J_{6.7a}$ 2.1, $J_{6.7b}$ 4.6, $J_{7a,7b}$ 12.4 Hz. Anal. Calcd for C₁₄H₂₀O₉: C, 50.60; H, 6.07. Found: C, 50.48; H, 6.21.

Methyl 2,6-anhydro-3-deoxy-D-gluco-heptonate (**10**).—To a cooled (-20 °C) solution of **9** (150 mg, 0.45 mmol) in MeOH (10 mL) was added dropwise Me₃SiCl (0.5 mL). The resulting mixture was stirred at room temperature overnight (TLC, 9:1 CH₂Cl₂-MeOH). The mixture was then neutralised with NaHCO₃ and filtered through a short column of silica gel to give **10** (85 mg, 95%) as white crystals; mp 139–141 °C, [α]_D +24.8° (c 0.9, MeOH); ¹H NMR (500 MHz, D₂O): δ 1.62 (dd, 1 H, H-3ax), 2.40 (ddd, 1 H, H-3eq), 3.31 (t, 1 H, H-5), 3.42 (ddd, 1 H, H-6), 3.76 (dd, 1 H, H-7b), 3.77 (m, 1 H, H-4), 3.80 (s, 3 H, COOMe), 3.92 (dd, 1 H, H-7a), 4.33 (dd, 1 H, H-2); $J_{2.3ax}$ 12.2, $J_{2.3eq}$ 2.3, $J_{3ax,3eq}$ 12.9, $J_{3ax,4}$ 12.3, $J_{3eq,4}$ 5.1, $J_{4.5}$ 9.2, $J_{5.6}$ 8.2, $J_{6.7a}$ 2.2, $J_{6.7b}$ 5.9, $J_{7a.7b}$ 12.4 Hz. Anal. Calcd for C₈H₁₄O₆: C, 46.60; H, 6.84. Found: C, 46.42; H, 7.04

Methyl 2,6-anhydro-3-deoxy-D-gluco-heptonate 7-(dihydrogen phosphate) (12).—The ester 10 (100 mg, 0.48 mmol) was dissolved in dry pyridine (3 mL) and freshly distilled diphenyl phosphorochloridate (0.1 mL, 0.50 mmol) was added at -20 °C under Ar. The reaction mixture was stirred until TLC (EtOAc) showed disappearance of 10 (20–30 min). Further processing was similar to that described for the preparation of 8, to give 12 (120 mg, 90%) as a white solid; [α]_D + 17.5° (c 1.20, MeOH); ¹H NMR (500 MHz, D₂O): δ 1.63 (dd, 1 H, H-3ax), 2.38 (ddd, 1 H, H-3eq), 3.40 (t, 1 H, H-5), 3.53 (bdd, 1 H, H-6), 3.76 (ddd, 1 H, H-4), 3.78 (s, 3 H, COOMe), 4.08 (m, 1 H, H-7a), 4.17 (ddd, 1 H, H-7b), 4.33 (dd, 1 H, H-2); $J_{2,3ax}$ 12.2, $J_{2,3eq}$ 2.2, $J_{3ax,3eq}$ 12.9, $J_{3ax,4}$ 12.3, $J_{3eq,4}$ 5.1, $J_{4,5}$ 9.2, $J_{5,6}$ 9.5, $J_{6,7a}$ 1.8, $J_{6,7b}$ 5.1, $J_{7a,7b}$ 11.5, $J_{7a,P}$ 5.5 Hz. HR-MS (LSIMS): Calcd for C₈H₁₅O₉P [M + Na]: m/z 309.0351. Found: m/z 309.0341.

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